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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=295 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.007 \AA$
Disorder in main residue
$R$ factor $=0.057$
$w R$ factor $=0.124$
Data-to-parameter ratio $=10.7$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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## Aqua(1,10-phenanthroline)[1,1,1-trifluoro-3-(2-thenoyl)acetonato]copper(II) (1,10-phen-anthroline)(5-sulfonatosalicylato)[1,1,1-trifluoro-3-(2-thenoyl)acetonato]cuprate(II)

The structure of the title complex is built up from a $\left[\mathrm{Cu}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{~F}_{3} \mathrm{O}_{2} \mathrm{~S}\right)\left(\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2}\right)\left(\mathrm{H}_{2} \mathrm{O}\right)\right]^{+}$cation and a $\left[\mathrm{Cu}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{~F}_{3} \mathrm{O}_{2} \mathrm{~S}\right)\left(\mathrm{C}_{7} \mathrm{H}_{4} \mathrm{O}_{6} \mathrm{~S}\right)\left(\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2}\right)\right]^{-}$anion. Both the cation and anion contain $\mathrm{Cu}^{\mathrm{II}}$ atoms having a squarepyramidal geometry. Hydrogen bonds between cations and anions generate a one-dimensional chain and stabilize the crystal packing.

## Comment

The study of cation-anion 5 -sulfosalicylate metal complexes has recently been greatly developed in our laboratory. These compounds exhibit interesting assembly topologies (Chen et al., 2005; Fan \& Zhu, 2005a,b; Fan et al., 2005) in complexes containing 5 -sulfosalicylate as the only anionic ligand. Here, we present the results of a combination of 5-sulfosalicylic acid and 2-thenoyltrifluoroacetone, giving the title cation-anion complex containing two different anionic ligands, (I).

(I)

In the structure of (I), both cation and anion contain a $\mathrm{Cu}^{\mathrm{II}}$ atom with a square-pyramidal geometry, in which the basal plane is defined by two N atoms from one 1,10-phenanthroline ligand and two O atoms from one 2-thenoyltrifluoroacetonate (ttf) ligand. The apical position in the cation is occupied by the water molecule, whereas in the anion it is occupied by the dianionic 5 -sulfosalicylate ligand (Figs. 1 and 2). In both species, the $\mathrm{Cu}-\mathrm{N}$ bond lengths are close to those of other carboxylate copper complexes containing N donor ligands (Lemoine et al., 2002; Zhu \& Kitagawa, 2002), and the Cu $\mathrm{O}(\mathrm{ttf})$ bond lengths are also similar to those of ttf -copper complexes (Yang et al., 2000, 2001).

It is worth noting that the coordinated water molecule in the cation forms hydrogen bonds with the phenolic atom O3 (symmetry code: $x,-1+y, z$ ) and the sulfonate atom O6 (symmetry code: $-1+x,-1+y, z$ ) from two adjacent anions, thereby creating a hydrogen-bonded chain (Fig. 3 and Table 1).


Figure 1
A view of the cation of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the $40 \%$ probability level. The minor disorder component of the $\mathrm{CF}_{3}$ group is not shown.

## Experimental

A solution of $\mathrm{CuSO}_{4} \cdot 5 \mathrm{H}_{2} \mathrm{O}(0.051 \mathrm{~g}, 0.20 \mathrm{mmol})$ and 5 -sulfosalicylic acid dihydrate $(0.051 \mathrm{~g}, 0.20 \mathrm{mmol})$ in water $(15 \mathrm{ml})$ was added to a solution of 2-thenoyltrifluoroacetone ( $0.044 \mathrm{~g}, 0.16 \mathrm{mmol}$ ) and $1,10-$ phenanthroline $(0.040 \mathrm{~g}, 0.20 \mathrm{mmol})$ in $N, N$-dimethylformamide $(10 \mathrm{ml})$ with stirring. The resulting solution was then set aside and allowed to evaporate. After three weeks, green block-shaped crystals of (I) were obtained and collected by suction filtration.

## Crystal data

$\left[\mathrm{Cu}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{~F}_{3} \mathrm{O}_{2} \mathrm{~S}\right)\left(\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2}\right)\left(\mathrm{H}_{2} \mathrm{O}\right)\right]-$
$\left[\mathrm{Cu}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{~F}_{3} \mathrm{O}_{2} \mathrm{~S}\right)\left(\mathrm{C}_{7} \mathrm{H}_{4} \mathrm{O}_{6} \mathrm{~S}\right)\right.$ -
$\left.\left(\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2}\right)\right]$

## $M_{r}=1164.01$

Monoclinic, $P 2_{1} / c$
$a=10.1963$ (5) $\AA$
$b=14.0035$ (7) $\AA$
$c=32.1357$ (17) $\AA$
$\beta=98.508$ (1) ${ }^{\circ}$
$V=4538.0$ ( 4 ) $\AA^{3}$

## Data collection

Bruker SMART APEX areadetector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan
(SADABS; Bruker, 2002)
$T_{\text {min }}=0.826, T_{\text {max }}=0.912$
23617 measured reflections

$$
\begin{aligned}
& Z=4 \\
& D_{x}=1.704 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation }
\end{aligned}
$$

Cell parameters from 4606
reflections
$\theta=2.4-25.5^{\circ}$
$\mu=1.17 \mathrm{~mm}^{-1}$
$T=295$ (2) K
Block, green
$0.17 \times 0.10 \times 0.08 \mathrm{~mm}$

> 7984 independent reflections
> 6409 reflections with $I>2 \sigma(I)$
> $R_{\text {int }}=0.041$
> $\theta_{\max }=25.0^{\circ}$
> $h=-11 \rightarrow 12$
> $k=-16 \rightarrow 16$
> $l=-35 \rightarrow 38$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.057$
$w R\left(F^{2}\right)=0.124$
$S=1.13$
7984 reflections
746 parameters
H atoms treated by a mixture of independent and constrained refinement


Figure 2
A view of the anion of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the $40 \%$ probability level. The minor disorder components of the sulfonate and $\mathrm{CF}_{3}$ groups are not shown.

Table 1
Hydrogen-bond geometry ( $\AA,{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O} 3-\mathrm{H} 3 \cdots \mathrm{O} 2$ | 0.82 | 1.73 | 2.466 (4) | 148 |
| $\mathrm{O} 11-\mathrm{H} 11 A \cdots \mathrm{O}^{\text {i }}$ | 0.84 (1) | 2.11 (1) | 2.947 (4) | 172 (4) |
| $\mathrm{O} 11-\mathrm{H} 11 B \cdots \mathrm{O} 6^{\text {, }{ }^{\text {ii }}}$ | 0.85 (1) | 1.87 (2) | 2.695 (16) | 166 (4) |
| $\mathrm{O} 11-\mathrm{H} 11 B \cdots \mathrm{O}^{\text {ii }}$ | 0.85 (1) | 1.89 (2) | 2.735 (15) | 177 (5) |

Symmetry codes: (i) $x, y-1, z$; (ii) $x-1, y-1, z$.
The H atoms attached to C or O (hydroxyl) atoms were placed in idealized positions and refined as riding on their parent atoms, with $\mathrm{C}-\mathrm{H}=0.93 \AA$ and $\mathrm{O}-\mathrm{H}=0.85 \AA$, and with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C}, \mathrm{O})$. Water H atoms were located in difference Fourier maps and were refined with $\mathrm{O}-\mathrm{H}$ and $\mathrm{H} \cdots \mathrm{H}$ restraints of 0.85 (1) and 1.39 (2) $\AA$,


Figure 3
A view of the hydrogen-bonding chain of (I). Hydrogen bonds are drawn as dashed lines. H atoms, 1,10 -phenanthroline ligands, and minor disorder components have been omitted for clarity.

## metal-organic papers

respectively, and with $U_{\text {iso }}(\mathrm{H})=1.5 U_{\text {eq }}(\mathrm{O})$. The sulfonate group and the $\mathrm{CF}_{3}$ groups are disordered over two positions each and their occupancies are 0.513 (14):0.487 (14) for atoms $\mathrm{O} 4 / \mathrm{O} 5 / \mathrm{O} 6$ and $\mathrm{O}^{\prime} /$ $\mathrm{O}^{\prime} / \mathrm{O}^{\prime}, 0.57$ (2):0.43 (2) for atoms $\mathrm{F} 1 / \mathrm{F} 2 / \mathrm{F} 3$ and $\mathrm{F}^{\prime} / \mathrm{F} 2^{\prime} / \mathrm{F} 3^{\prime}$, and 0.584 (19):0.416 (19) for atoms F4/F5/F6 and $\mathrm{F} 4^{\prime} / \mathrm{F} 5^{\prime} / \mathrm{F} 6^{\prime}$, respectively. The anisotropic displacement parameters for these disordered groups were refined using similarity restraints.

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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