metal-organic papers

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Key indicators

Single-crystal X-ray study T = 295 K Mean σ (C–C) = 0.007 Å Disorder in main residue R factor = 0.057 wR factor = 0.124 Data-to-parameter ratio = 10.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Aqua(1,10-phenanthroline)[1,1,1-trifluoro-3-(2-thenoyl)acetonato]copper(II) (1,10-phenanthroline)(5-sulfonatosalicylato)[1,1,1trifluoro-3-(2-thenoyl)acetonato]cuprate(II)

The structure of the title complex is built up from a $[Cu(C_8H_4F_3O_2S)(C_{12}H_8N_2)(H_2O)]^+$ cation and a $[Cu(C_8H_4F_3O_2S)(C_7H_4O_6S)(C_{12}H_8N_2)]^-$ anion. Both the cation and anion contain Cu^{II} atoms having a square-pyramidal geometry. Hydrogen bonds between cations and anions generate a one-dimensional chain and stabilize the crystal packing.

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Comment

The study of cation–anion 5-sulfosalicylate metal complexes has recently been greatly developed in our laboratory. These compounds exhibit interesting assembly topologies (Chen *et al.*, 2005; Fan & Zhu, 2005*a,b*; Fan *et al.*, 2005) in complexes containing 5-sulfosalicylate as the only anionic ligand. Here, we present the results of a combination of 5-sulfosalicylic acid and 2-thenoyltrifluoroacetone, giving the title cation–anion complex containing two different anionic ligands, (I).



In the structure of (I), both cation and anion contain a Cu^{II} atom with a square-pyramidal geometry, in which the basal plane is defined by two N atoms from one 1,10-phenanthroline ligand and two O atoms from one 2-thenoyltrifluoroacetonate (ttf) ligand. The apical position in the cation is occupied by the water molecule, whereas in the anion it is occupied by the dianionic 5-sulfosalicylate ligand (Figs. 1 and 2). In both species, the Cu–N bond lengths are close to those of other carboxylate copper complexes containing N donor ligands (Lemoine *et al.*, 2002; Zhu & Kitagawa, 2002), and the Cu–O(ttf) bond lengths are also similar to those of ttf–copper complexes (Yang *et al.*, 2000, 2001).

It is worth noting that the coordinated water molecule in the cation forms hydrogen bonds with the phenolic atom O3 (symmetry code: x, -1 + y, z) and the sulfonate atom O6 (symmetry code: -1 + x, -1 + y, z) from two adjacent anions, thereby creating a hydrogen-bonded chain (Fig. 3 and Table 1).

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Figure 1

A view of the cation of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 40% probability level. The minor disorder component of the CF₃ group is not shown.

Experimental

A solution of CuSO₄·5H₂O (0.051 g, 0.20 mmol) and 5-sulfosalicylic acid dihydrate (0.051 g, 0.20 mmol) in water (15 ml) was added to a solution of 2-thenoyltrifluoroacetone (0.044 g, 0.16 mmol) and 1,10phenanthroline (0.040 g, 0.20 mmol) in N,N-dimethylformamide (10 ml) with stirring. The resulting solution was then set aside and allowed to evaporate. After three weeks, green block-shaped crystals of (I) were obtained and collected by suction filtration.

Crystal data

$[Cu(C_8H_4F_3O_2S)(C_{12}H_8N_2)(H_2O)]$ -	Z = 4
$[Cu(C_8H_4F_3O_2S)(C_7H_4O_6S)-$	$D_x = 1.704 \text{ Mg m}^{-3}$
$(C_{12}H_8N_2)]$	Mo $K\alpha$ radiation
$M_r = 1164.01$	Cell parameters from 4
Monoclinic, $P2_1/c$	reflections
a = 10.1963 (5) Å	$\theta = 2.4 - 25.5^{\circ}$
b = 14.0035 (7) Å	$\mu = 1.17 \text{ mm}^{-1}$
c = 32.1357 (17) Å	T = 295 (2) K
$\beta = 98.508 \ (1)^{\circ}$	Block, green
V = 4538.0 (4) Å ³	$0.17 \times 0.10 \times 0.08 \text{ mm}$
Data collection	

Bruker SMART APEX areadetector diffractometer φ and ω scans Absorption correction: multi-scan (SADABS; Bruker, 2002) $T_{\min} = 0.826, T_{\max} = 0.912$ 23617 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.057$ wR(F²) = 0.124 S = 1.137984 reflections 746 parameters H atoms treated by a mixture of independent and constrained refinement

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7984 independent reflections 6409 reflections with $I > 2\sigma(I)$ $R_{\rm int}=0.041$ $\theta_{\rm max} = 25.0^\circ$ $h = -11 \rightarrow 12$ $k = -16 \rightarrow 16$ $l = -35 \rightarrow 38$

 $w = 1/[\sigma^2(F_0^2) + (0.0483P)^2]$ + 1.9253P] where $P = (F_0^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\rm max} = 0.003$ $\Delta \rho_{\rm max} = 0.44$ e Å⁻³ $\Delta \rho_{\rm min} = -0.34 \text{ e} \text{ Å}^{-3}$



Figure 2

A view of the anion of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 40% probability level. The minor disorder components of the sulfonate and CF₃ groups are not shown.

Table 1 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\begin{array}{c} 03 - H3 \cdots 02 \\ 011 - H11A \cdots 03^{i} \\ 011 - H11B \cdots 06^{iii} \\ 011 - H11B \cdots 06^{ii} \end{array}$	0.82 0.84 (1) 0.85 (1) 0.85 (1)	1.73 2.11 (1) 1.87 (2) 1.89 (2)	2.466 (4) 2.947 (4) 2.695 (16) 2.735 (15)	148 172 (4) 166 (4) 177 (5)

Symmetry codes: (i) x, y - 1, z; (ii) x - 1, y - 1, z.

The H atoms attached to C or O(hydroxyl) atoms were placed in idealized positions and refined as riding on their parent atoms, with C-H = 0.93 Å and O-H = 0.85 Å, and with $U_{iso}(H) = 1.2U_{eq}(C,O)$. Water H atoms were located in difference Fourier maps and were refined with O-H and H···H restraints of 0.85 (1) and 1.39 (2) Å,



Figure 3

A view of the hydrogen-bonding chain of (I). Hydrogen bonds are drawn as dashed lines. H atoms, 1,10-phenanthroline ligands, and minor disorder components have been omitted for clarity.

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respectively, and with $U_{\rm iso}({\rm H}) = 1.5U_{\rm eq}({\rm O})$. The sulfonate group and the CF₃ groups are disordered over two positions each and their occupancies are 0.513 (14):0.487 (14) for atoms O4/O5/O6 and O4'/O5'/O6', 0.57 (2):0.43 (2) for atoms F1/F2/F3 and F1'/F2'/F3', and 0.584 (19):0.416 (19) for atoms F4/F5/F6 and F4'/F5'/F6', respectively. The anisotropic displacement parameters for these disordered groups were refined using similarity restraints.

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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